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## C-27 AND C-3 GLUCOSYLATION OF DIOSGENIN BY CELL SUSPENSION CULTURES OF *COSTUS SPECIOSUS*

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3-O- $[\beta$ -D-glucopyranosyl-(1''  $\rightarrow$  2')- $\beta$ -D-glucopyranosyl], 27-O- $\beta$ -D-glucopyranosyl-(25R)-spirost-5-ene-3 $\beta$ ,27-diol was isolated from cell suspension cultures of *Costus speciosus*, following incubation with diosgenin, and its structure was elucidated using a combination of one- and two-dimensional <sup>1</sup>H and <sup>13</sup>C NMR spectral data, and positive and negative ion ESMS spectral data.

**Keywords:** *Costus speciosus*; Cell suspension cultures; Biotransformation; Glucosylation; Diosgenin; 3-O- $[\beta$ -D-glucopyranosyl-(1''  $\rightarrow$  2')- $\beta$ -D-glucopyranosyl]; 27-O- $\beta$ -D-glucopyranosyl-(25R)-spirost-5-ene-3 $\beta$ ,27-diol

### INTRODUCTION

Various plant cell cultures are capable of glucosylating a variety of exogenously supplied substrates [1]. For example, the glucosylation of simple phenols by various cell suspension cultures has been reported by Umetami, Tanaka and Tabata [2]. Cell suspension cultures of *Salix matsudana* transform salicyl alcohol into salicin and isosalicin, while salicylic acid was converted into salicylic acid-2-O- $\beta$ -D-glucopyranoside and

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salicylic acid-7-O- $\beta$ -D-glucopyranoside [3]. In previous work, we reported the bioconversion of salicyl alcohol into salicin, salicylamide into salicylamide 2-O- $\beta$ -D-glucopyranoside, *p*-aminobenzoic acid into *p*-aminobenzoic acid-7-O- $\beta$ -D-glucopyranosyl ester, and *o*-aminobenzoic acid into *o*-aminobenzoic acid-7-O- $\beta$ -D-glucopyranosyl ester and *o*-aminobenzoic acid-7-O- $\beta$ -D-( $\beta$ -D-1',6-O-glucopyranosyl)-glucopyranosyl ester by cell suspension cultures of *Solanum mammosum* [4–6]. Paczkowski and Wojciechowski [7] reported the glucosylation of diosgenin and solasodine into its mono-glucoside by soluble glycosyltransferase(s) from *Solanum melongena* leaves, while Inoue *et al.* [8] have demonstrated the transformation of a furostanol glycoside (protogracillin) into a spirostanol glycoside (gracillin) by  $\beta$ -glucosidase in *Costus speciosus* rhizomes. These studies showed that diosgenin glycoside can be produced by both glycosyltransferase and glucosidase in plant cells. The fungus *Cunninghamella elegans* has been reported to transform diosgenin into its di- and tri-hydroxy derivatives [9].

The callus and cell suspension cultures of *C. speciosus* contain cholesterol, campesterol, stigmasterol and  $\beta$ -sitosterol, while diosgenin (**1**) and other steroidal sapogenins have not been detected. Diosgenin accumulates only in shoot cultures of *C. speciosus* [10].

It is well known that the accumulation of secondary metabolites depends on the relative rates of biosynthesis and catabolism; accumulation can occur when the rate of biosynthesis is higher than the rate of catabolism [11]. The absence of **1** in the callus and suspension cultures of *C. speciosus* may be due to its conversion, immediately after its biosynthesis. In order to determine whether **1** can be transformed to other substances, it was incubated in cell suspension cultures of *C. speciosus*. Another aim of the experiment was to determine whether diosgenin glycoside(s) can be formed by direct glycosylation of the aglycone **1**.

We now report the biotransformation of diosgenin (**1**) by cell suspension cultures of *C. speciosus* into a new C-27 glucoside, 3-O- $[\beta$ -D-glucopyranosyl-(1''  $\rightarrow$  2')- $\beta$ -D-glucopyranosyl], 27-O- $\beta$ -D-glucopyranosyl-(25*R*)-spirost-5-ene-3 $\beta$ ,27-diol (**2**), and its structure elucidation using a combination of one- and two-dimensional  $^1\text{H}$  and  $^{13}\text{C}$  NMR, and positive and negative ion ESMS spectral data.

## RESULTS AND DISCUSSION

Incubation of cell suspension cultures of *Costus speciosus* with **1** (75 mg l<sup>-1</sup>), followed by isolation, purification by column chromatography and

preparative TLC, afforded metabolite **2** ( $R_f$  0.19). Spots of compounds **1** and **2** showed identical visible absorbance reflectance spectra after reaction with anisaldehyde-sulphuric acid reagent. Control experiments showed that in the absence of cells, substrate **1** ( $R_f$  0.81) remained unchanged in the culture medium, and that metabolite **2** was only produced when cell suspension cultures of *C. speciosus* were present in the medium. Although the PCV (packed cell volume) and GI (growth index) were decreased (*ca.* 40%), cell suspension cultures of *C. speciosus* seemed to be resistant to inoculation with **1** ( $75 \text{ mg l}^{-1}$ ).

The positive ion electron spray mass spectrum (ESMS) of **2**, determined in a cation-assisted matrix, displayed pseudomolecular ions at  $m/z$  917 ( $M+H$ )<sup>+</sup>, 934 ( $M+NH_4$ )<sup>+</sup>, 939 ( $M+Na$ )<sup>+</sup>, and 955 ( $M+K$ )<sup>+</sup>. The negative ion ESMS determined in a  $Cl^-$  assisted matrix, showed pseudomolecular ions at  $m/z$  915 ( $M-H$ )<sup>-</sup>, 951 ( $M+^{35}Cl$ )<sup>-</sup> and 953 ( $M+^{37}Cl$ )<sup>-</sup>. These data indicated that metabolite **2** had a molecular weight of 916 daltons demonstrating **2** to be a trihexoside analogue of **1**.

The  $^{13}C$  NMR spectrum of **2** exhibited 45 carbon signals, including a quaternary acetal-type resonance at  $\delta$  109.6 (C-22), two alkene resonances at  $\delta$  141.1 (C-5) and 121.6 (C-6), and three methyl group resonances at  $\delta$  16.4 (C-18), 19.5 (C-19), and 15.0 (C-21), all of which corresponded closely to those observed for diosgenin; however, a fourth methyl group resonance was not observed. The  $^{13}C$  NMR spectrum also displayed three anomeric glucosyl signals at  $\delta$  101.5 (C-1'), 106.7 (C-1'') and 105.1 (C-1''').

The  $^1H$  NMR spectrum of **2** displayed signals attributable to two tertiary methyl groups at  $\delta$  0.82 (H-18) and 1.01 (H-19), while only a single secondary methyl group was observed at  $\delta$  1.10 (*d*,  $J=6.9$  Hz). Three anomeric  $\beta$ -D-glucopyranosyl proton resonances were also observed at  $\delta$  5.06 (H-1', *d*,  $J=7.6$  Hz), 5.27 (H-1'', *d*,  $J=7.7$  Hz) and 4.77 (H-1''', *d*,  $J=7.7$  Hz).

The HMBC spectrum of **2** displayed correlations between the anomeric glucosyl protons at  $\delta$  5.06 (H-1') with C-3 ( $\delta$  79.3) and  $\delta$  5.27 (H-1'') with C-2' ( $\delta$  84.8), while H-2' ( $\delta$  4.14) exhibited correlations with C-1' ( $\delta$  101.5), C-1'' ( $\delta$  106.7) and C-3' ( $\delta$  77.9). These data indicated that the inner  $\beta$ -D-glucopyranosyl residue was attached to C-3, and that the pair of  $\beta$ -D-glucopyranosyl residues were mutually 1, 2-linked. Comparison with the  $^{13}C$  NMR spectral assignments determined for the sugar moieties of diosgenin-3-O- $\beta$ -D-glucopyranosyl-(1''  $\rightarrow$  3')- $\beta$ -D-glucopyranoside (glucoside **3**) [12] showed that the C-2' resonance of **2** exhibited a downfield shift of 10.4 ppm, indicative of  $\alpha$ -glucosylation [13]. Correlations observed in the ROESY spectrum of **2** also established the presence of a 1, 2-glucosidic linkage. In particular, correlations were observed between H-1' ( $\delta$  5.06) and H-3 ( $\delta$  3.84) and between H-1'' ( $\delta$  5.27) and H-2' ( $\delta$  4.14).

One- and two-dimensional NMR spectral data (including COSY, TOCSY, ROESY, HMBC and HSQC spectra) showed that the third glucopyranosyl residue was attached to C-27. TOCSY and COSY correlations identified the resonances and connectivities of other protons associated with each of the glycosyl residues, while the HSQC spectrum identified the corresponding  $^{13}\text{C}$  resonances. The HMBC spectrum of **2** indicated correlations between H-27<sub>A</sub> ( $\delta$  3.46) and C-1''' ( $\delta$  105.1), C-26 ( $\delta$  63.7), C-25 ( $\delta$  36.7), C-24 ( $\delta$  24.0); H-27<sub>B</sub> ( $\delta$  3.93) and C-26, C-25, C-24; and H-1''' ( $\delta$  4.77) and C-27 ( $\delta$  72.0). The attachment of the third glucopyranosyl residue to C-27 was also confirmed by the ROESY spectrum of **2**, which showed a correlation between H-1''' and H-27<sub>A</sub>.

Consequently the structure of **2** was determined to be 3-O-[ $\beta$ -D-glucopyranosyl-(1'' - 2')- $\beta$ -D-glucopyranosyl], 27-O- $\beta$ -D-glucopyranosyl-(25*R*)-spirost-5-ene-3 $\beta$ ,27-diol.

To our knowledge this is the first report of the C-27 glucosylation of diosgenin by cell suspension cultures of *C. speciosus*. This is also the first report of the isolation of 3-O-[ $\beta$ -D-glucopyranosyl-(1'' - 2')- $\beta$ -D-glucopyranosyl], 27-O- $\beta$ -D-glucopyranosyl-(25*R*)-spirost-5-ene-3 $\beta$ ,27-diol from a natural source. This work showed that *C. speciosus* cells can biosynthesize diosgenin glycoside by glucosylation of the aglycone, in addition to the bioconversion of the furostanol glycoside as previously reported [8].

It seems likely that cell suspension cultures of *C. speciosus* hydroxylate diosgenin at C-27 prior to glucosylation. We are investigating the possibility of isolating 27-hydroxydiosgenin, the presumed intermediate biotransformation product, and endeavour to characterize the enzymes that are responsible for the bioconversion.

## EXPERIMENTAL SECTION

### General Experimental Procedures

NMR spectra were recorded at 400.13 ( $^1\text{H}$ ) and 100.62 MHz ( $^{13}\text{C}$ ) using an inverse 5 mm probehead installed in a Bruker DRX 400 spectrometer. Gradient selection was utilized in HMBC and HSQC experiments. Chemical shifts ( $\delta$  ppm) are reported relative to solvent peaks observed for pyridine-*d*<sub>5</sub> ( $^1\text{H}$  = 8.70 ppm, low field signal;  $^{13}\text{C}$  = 123.5 ppm, high field signal). Coupling constants are reported to a precision of  $\pm 0.2$  Hz.  $^{13}\text{C}$  NMR signal multiplicities (*d*, *t* or *q*; *s* suppressed) were determined using the DEPT sequence with a  $135^\circ$  detection pulse. Two-dimensional COSY and HMBC

(80 msec mixing time) spectra were determined in absolute value mode, while TOCSY, ROESY (250 msec spin lock time) and HSQC spectra were determined in phase-sensitive mode.

Positive and negative ion electrospray mass spectra (ESMS) were obtained using a Fisons VG Platform II instrument. Samples were introduced into the spectrometer using  $\text{CH}_3\text{CN}:\text{H}_2\text{O}$  (1:1) as solvent.

The visible absorbance reflectance spectra of the spots of substrate **1** and metabolite **2** were determined using a Shimadzu TLC Scanner CS 930. Stationary phase: silica gel  $\text{F}_{254}$  precoated plate (E. Merck); mobile phase:  $\text{CHCl}_3:\text{MeOH}:\text{H}_2\text{O}$  (40:16:3); detection was performed with anisaldehyde-sulphuric acid reagent (100°C, 5 min).

### Cell Suspension Culture and Biotransformation Conditions

Cell suspension cultures were initiated from callus cultures of *C. speciosus* (code F8), as previously reported [10]. The calli were cultivated in 300 ml Erlenmeyer flasks containing 50 ml of modified Murashige and Skoog medium [14] supplemented with sucrose ( $30\text{ g l}^{-1}$ ), kinetin ( $2\text{ mg l}^{-1}$ ), 2,4-dichloro phenoxyacetic acid ( $0.5\text{ mg l}^{-1}$ ) and casein hydrolysate ( $1\text{ g l}^{-1}$ ) on a gyrotary shaker (120 rpm) at  $25 \pm 1^\circ$  under continuous light (ca. 1500 lux). Biotransformation experiments were typically performed by inoculating cells (6–7 g fresh weight) into liquid medium (50 ml) containing **1** ( $75\text{ mg l}^{-1}$ ) and incubated for 12 days. After 12 days cultures were harvested, followed by PCV (packed cell volume) and GI (growth index) determination [10], filtered, weighed, oven dried at  $40^\circ$  (until their water content was ca. 2%), and powdered.

### Biotransformation of **1** and Isolation of **2**

Incubation of diosgenin **1** ( $75\text{ mg l}^{-1}$ ), and with the *C. speciosus* cell cultures as described above, produced powdered biomass (23.2 g), which was refluxed for 1 h with  $\text{CHCl}_3$  (4 times); the residues were then refluxed (4 times) with MeOH for 2 h. Concentration of the combined MeOH extracts under reduced pressure using a rotatory evaporator afforded a semi-solid brown residue (1.93 g), which was submitted to column chromatography on silica gel 60 (E. Merck, 70–230 mesh) using  $\text{CHCl}_3:\text{MeOH}:\text{H}_2\text{O}$  (40:16:3) as eluent [12]. Purification of the metabolite fraction by preparative TLC (silica gel 60  $\text{F}_{254}$  precoated plate, Merck; 0.25 mm layer) using  $\text{CHCl}_3:\text{MeOH}:\text{H}_2\text{O}$  (40:16:3) as developing solvent, yielded metabolite **2** (10 mg).

TABLE 1  $^1\text{H}$  and  $^{13}\text{C}$  NMR chemical shifts for glucosides **2** and **3** (ppm in pyridine- $d_5$ )<sup>a</sup>

<i>Position</i>	<b>2</b> $^1\text{H}$	<b>2</b> $^{13}\text{C}$	<b>3</b> <sup>b</sup> $^{13}\text{C}$
1 <sub>A</sub>	0.93	37.5	37.4
1 <sub>B</sub>	1.68		
2 <sub>A</sub>	1.80	30.3	30.2
2 <sub>B</sub>	2.14		
3	3.84	79.3	78.8
4 <sub>A</sub>	2.65	39.3	39.3
4 <sub>B</sub>	2.82		
5		141.1	140.8
6	5.34	121.6	121.8
7 <sub>A</sub>	1.44	32.3	32.2
7 <sub>B</sub>	1.83		
8	1.51	31.7	31.6
9	0.84	50.3	50.2
10	...	37.1	37.0
11 <sub>A</sub>	1.32	21.2	21.1
11 <sub>B</sub>	1.45		
12 <sub>A</sub>	1.16	39.9	39.9
12 <sub>B</sub>	1.66		
13	...	40.5	40.5
14	1.01	56.7	56.6
15 <sub>A</sub>	1.98	32.2	31.8
15 <sub>B</sub>	1.40		
16	4.49	81.2	81.1
17	1.75	62.9	62.9
18	0.82	16.4	16.4
19	1.01	19.5	19.4
20	1.90	42.1	42.0
21	1.10 ( <i>d, J</i> = 6.9)	15.0	15.0
22	...	109.6	109.3
23 <sub>A</sub>	1.62	31.3	31.8
23 <sub>B</sub>	1.66		
24 <sub>A</sub>	1.62	24.0	29.3
24 <sub>B</sub>	1.66		
25	2.05	36.7	30.6
26 <sub>A</sub>	3.72	63.7	66.9
26 <sub>B</sub>	4.03		
27 <sub>A</sub>	3.46	72.0	17.3
27 <sub>B</sub>	3.93		
1'	5.06 ( <i>d, J</i> = 7.6)	101.5	102.1
2'	4.14	84.8	74.4
3'	4.35	77.9	88.9
4'	4.23	71.5	69.8
5'	3.88	78.2	78.1
6 <sub>A</sub>	4.34	62.7	62.5
6 <sub>B</sub>	4.50		
1''	5.27 ( <i>d, J</i> = 7.7)	106.7	106.1
2''	4.13	77.1	75.7
3''	4.24	78.1	78.3
4''	4.30	71.6	71.6
5''	3.98	78.6	78.3
6 <sub>A</sub> '	4.46	62.9	62.5
6 <sub>B</sub> '	4.57		

TABLE I (Continued)

Position	<b>2</b> $^1\text{H}$	<b>2</b> $^{13}\text{C}$	<b>3</b> <sup>b</sup> $^{13}\text{C}$
1''	4.77 ( <i>d</i> , <i>J</i> = 7.7)	105.1	—
2''	4.01	75.2	—
3''	4.23	78.6	—
4''	4.23	71.7	—
5''	3.95	78.8	—
6 <sub>A</sub> ''	4.38	62.9	—
6 <sub>B</sub> ''	4.56	—	—

<sup>a</sup>Signals were assigned by means of 2D NMR experiments. Coupling constants (*J* in Hz) are given in parentheses.

<sup>b</sup>Inoue *et al.*, 1995 [12].

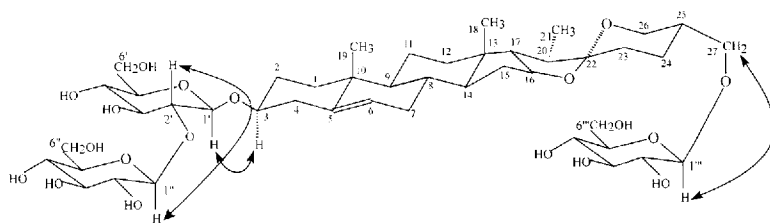


FIGURE 1 Chemical structure of metabolite **2** and selected ROESY correlations ( $\leftrightarrow$ ).

**Metabolite 2**, white amorphous solid; *in situ* visible absorbance reflectance,  $\lambda_{\text{max}}$  nm (silica gel F<sub>254</sub> precoated plate; anisaldehyde-sulphuric acid reagent): 428; positive ion ESMS (+60 V, Na<sup>+</sup>, K<sup>+</sup> and NH<sub>4</sub><sup>+</sup> assisted matrix) *m/z* (% rel. int.): 917 ([M+H]<sup>+</sup>, 100), 934 ([M+NH<sub>4</sub>]<sup>+</sup>, 40), 939 ([M+Na]<sup>+</sup>, 70), 955 ([M+K]<sup>+</sup>, 48); negative ion ESMS (−100 V, Cl<sup>−</sup> assisted matrix), *m/z* (% rel. int.): 915 ([M−H]<sup>−</sup>, 100), 951 ([M+<sup>35</sup>Cl]<sup>−</sup>, 23), 953 ([M+<sup>37</sup>Cl]<sup>−</sup>, 18); <sup>1</sup>H NMR and <sup>13</sup>C NMR spectral data see: Table I.

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